

# The coffee cup problem 2



How long does it take for the sucrose to diffuse to the top ?  
At least weeks, months for a big cup.

How long does it take to cool down to room temperature ?

What if diffusion in air is the only mechanism ?

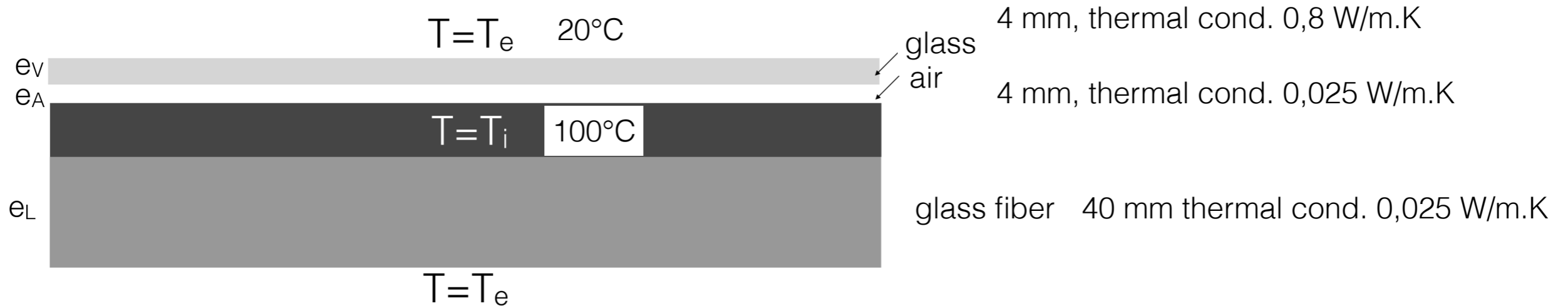
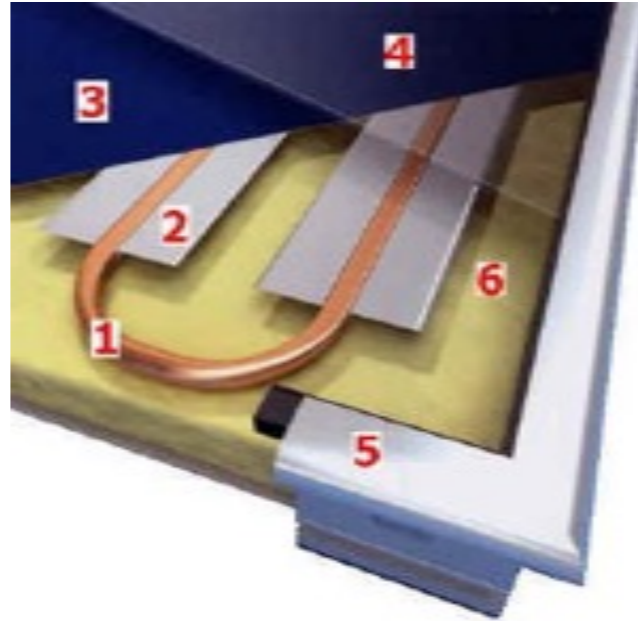
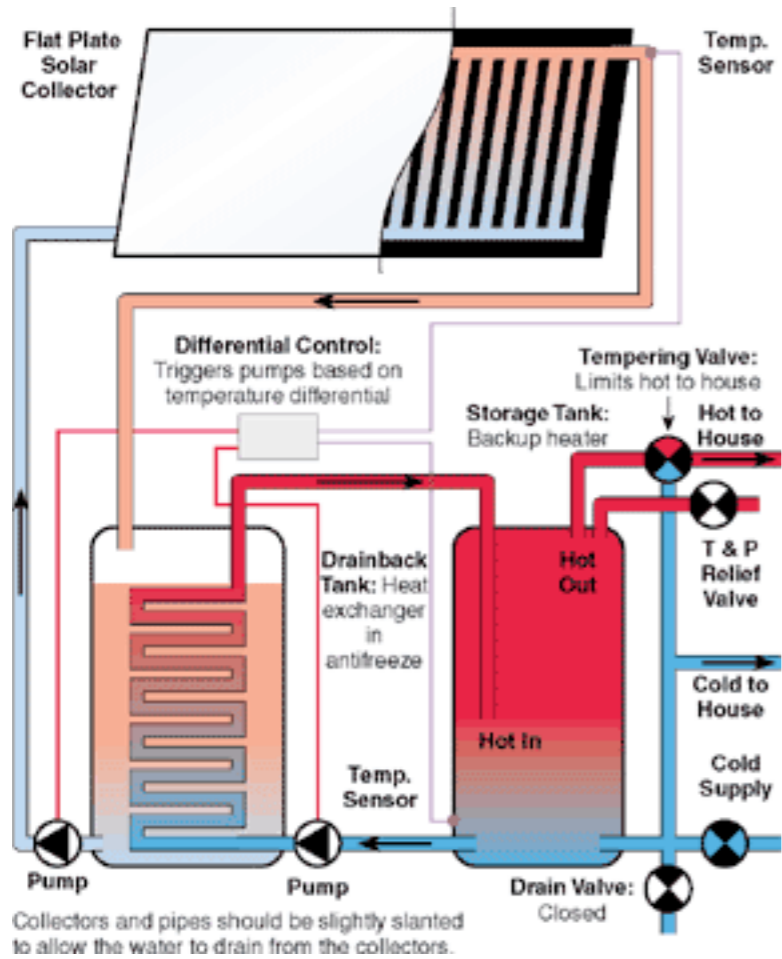
Physical properties of air :

density  $\rho = 1 \text{ kg/m}^3$

specific heat  $C_p = 1000 \text{ J/kg.K}$

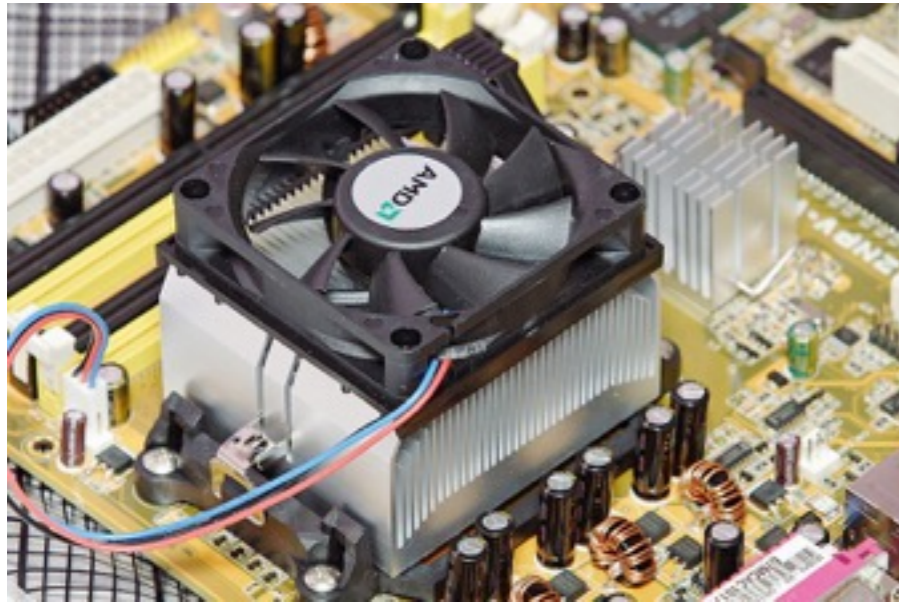
thermal conductivity  $\lambda = 0.025 \text{ W/m.K}$

# The solar heater problem I : how much heat do we lose from the collector ?



use the 1D steady state diffusion and the concept of thermal resistance to compute heat fluxes on both sides of the collector

# The microprocessor problem 1 and 2



Power to dissipate : 50 W

Typical dimension :  $L = 3\text{cm}$

Physical properties of air :

density  $1\text{ kg/m}^3$

specific heat  $1000\text{ J/kg.K}$

thermal conductivity  $0.025\text{ W/m.K}$

If we rely only on diffusion in air, how much power can we dissipate from the flat chip without fins?

how much power can we dissipate by adding a radiator with fins ?

is there an optimum length for fins given their thickness  $e$  and the heat transfer coefficient  $h$  between the fins and the surrounding air ?

Physical properties of aluminum (material of the fins) :

thermal conductivity  $240\text{ W/m.K}$

# The Maurice Herzog problem I



How long can you stand on top of Annapurna without gloves ? (no wind)

use the 1D bioheat equation to determine the temperature in Maurice's fingers (3mm of tissue between bone and skin) at steady state

steady state 1D diffusion + source term

blood flow rate per unit volume of tissue:  $V_b$   $10^{-4}$   $s^{-1}$

blood density:  $\rho_b$   $1000$   $kg/m^3$

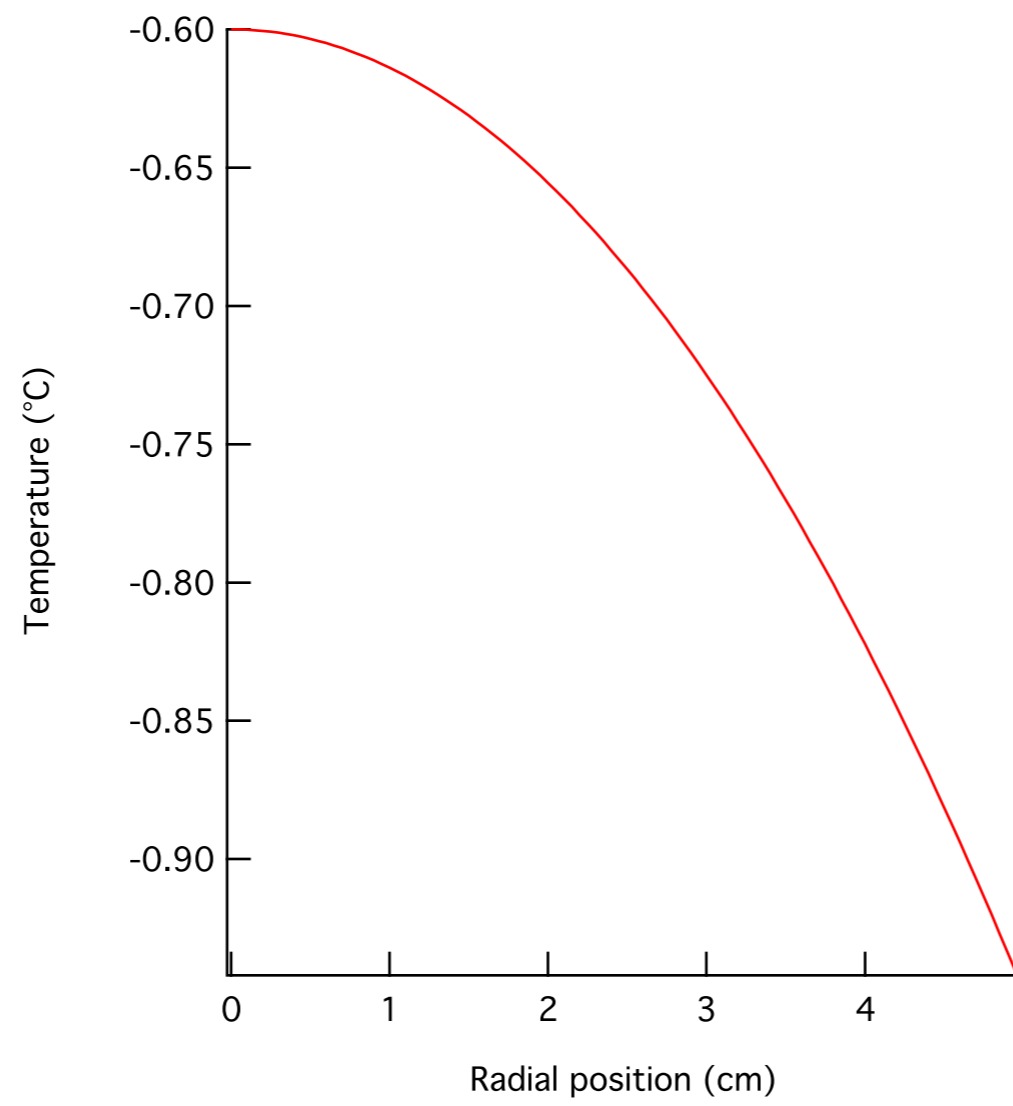
blood specific heat:  $C_b$   $3600$   $J/kg.K$

thermal conductivity of tissue:  $\lambda$   $0.5$   $W/m.K$

metabolic heat generation rate:  $q_m$   $700$   $W/m^3$

outside temperature:  $T_e$   $-20^\circ C$

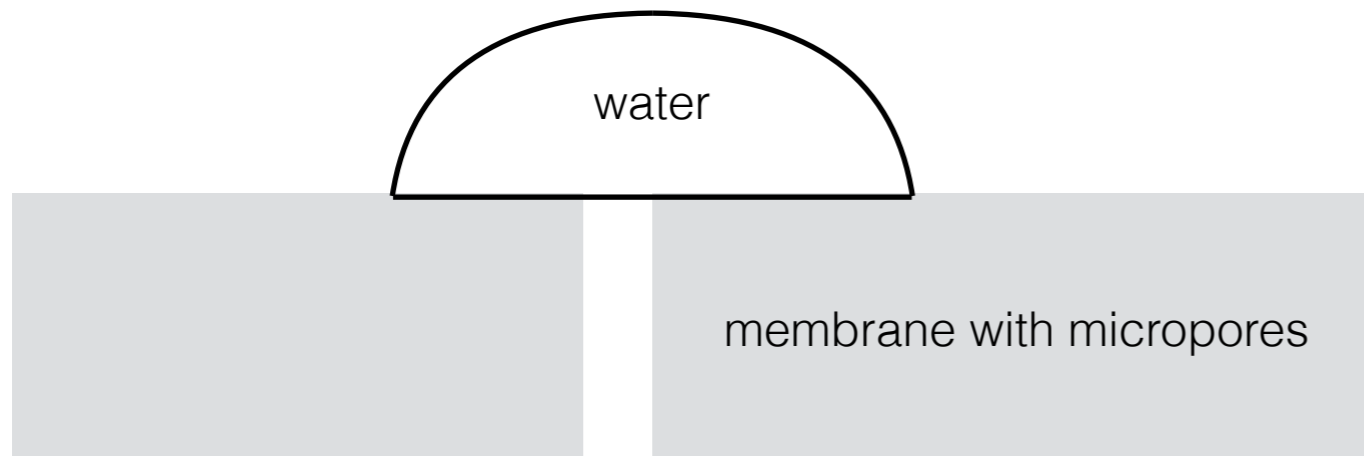
heat transfer coefficient with air:  $h$   $2$   $W/m^2.K$



# The Alex Thomson problem



how much water vapor is permeating through his clothing ?



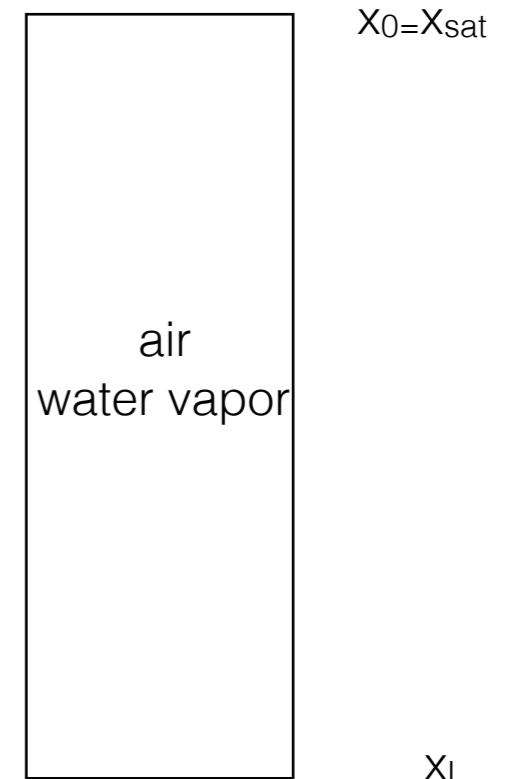
water vapor  $\phi\%$

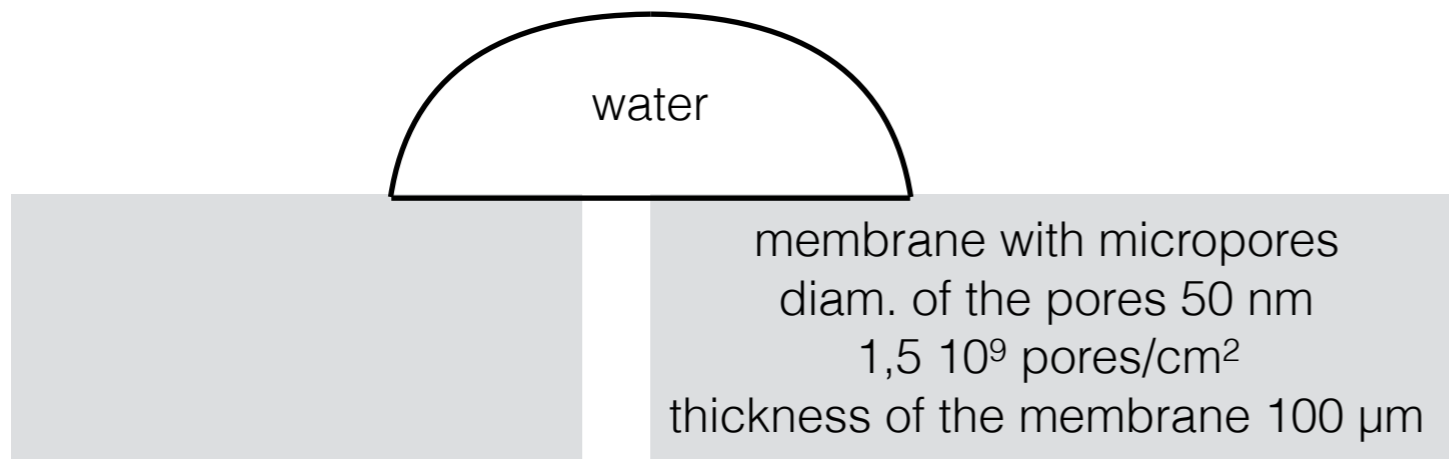
$$X_0 = X_{\text{sat}} = \frac{p_{\text{sat}}}{p}$$

$$X_L = \phi X_{\text{sat}} = \phi \frac{p_{\text{sat}}}{p}$$

$$p_{\text{sat}} = 0.03 \text{ bar at } 298 \text{ K}$$

$$D = 0.25 \cdot 10^{-4} \text{ m}^2/\text{s}$$





water vapor  $\phi\%$

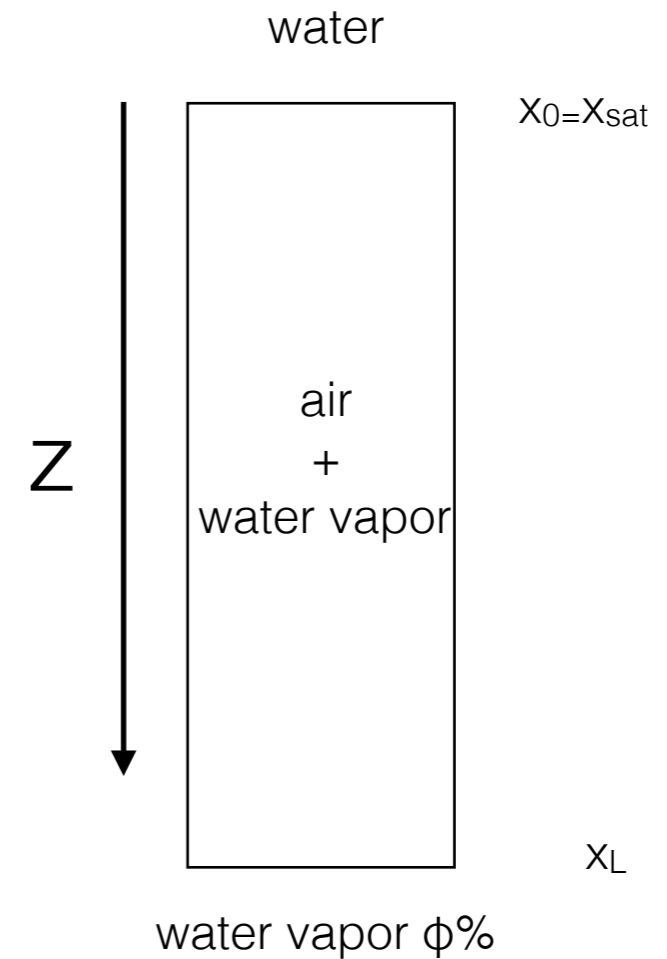
$$X_{A0} = X_{\text{sat}} = p_{\text{sat}}/p$$

$$X_{AL} = \phi X_{\text{sat}} = \phi p_{\text{sat}}/p$$

$$p_{\text{sat}} = 0.03 \text{ bar at } 298 \text{ K}$$

$$D_A = 0.25 \cdot 10^{-4} \text{ m}^2/\text{s}$$

the mixture A+B is treated as an ideal gaz mixture



Write the molar fluxes of vapor  $N_A$  and air  $N_B$  as a function of molar fractions  $x_A$  and  $x_B$

$$d N_A/dz = ? \quad d N_B/dz = ? \quad \text{value of } N_B \text{ at the gaz/water interface ?}$$